RESEARCHERID THOMSON REUTERS [Amit et al., 8(8), Aug 2023]



# INTERNATIONAL JOURNAL OF RECENT TECHNOLOGY SCIENCE & MANAGEMENT

"A STUDY OF NON- REGULAR EMISSION OF BLENDING FUEL IN ENGINE"

Amit Kumar<sup>1</sup>, Yogesh Kumar Tembhurne<sup>2</sup>

<sup>1</sup> M.Tech Scholar, Department of Mechanical Engineering, Bhabha university Bhopal, MP, India <sup>2</sup>Assistant Professor, Department of Mechanical Engineering, Bhabha university Bhopal, MP, India

#### ABSTRACT

The present work is an effort towards the review of literature of performance, combustion and emissions of unmodified compression ignition engine utilizing different alternative fuels. This study is established on the published work of eminent researchers between 2003 and 2020. The fast depletion of fossil fuels necessitates an alternative fuel and biodiesel may be deliberated as potential substitute fuel for CI engines to accomplish the global energy requirement. The various researchers and scientists conducted the experiments, using different alternative fuels viz. esters of edible oils, non-edible oils, alcohols and ethers and their mixing with diesel fuel. Use of either of these forms of resources beyond their rate of replacement is considered to be resource depletion. Resource depletion is most commonly used in reference to farming, fishing, mining, water usage, and consumption of fossil fuels.

Key Words: Fossil fuels, calorific value, biodiesel, compression ignition.

#### I. INTRODUCTION

Fossil fuels have a vital role in the evolution of industrial growth, transport sector, agriculture, and many individual needs. The reserves of petroleum fuel decrease every day in the world. Thus, most of the researchers are looking for alternative fuels. Renewable fuels are more attractive to the reserves of fossil fuels. These are facing many challenges, including environmental issues, feedstock, product commercialization, waste glycerol glut problem, and society's acceptance. Biodiesel will be used as a substitute for diesel to face present and future energy demand. Biodiesel is produced from edible and non-edible oils, which are eco-friendly, reliable, non-toxic, and biodegradable. Biodiesel emits low carbon and smoke emissions compared to diesel, which reduces global warming. Biodiesel has a higher density, pour point, viscosity, and molecular weight than diesel. Major biodiesel problems were low fuel atomization, low volatility, injector coking, and piston ring trapping, leading to incomplete combustion due to its viscosity and molecular weight. Biodiesel comprises a long chain of fatty acids containing 10-14 percent oxygen by weight and does not contain sulfur and aromatics. Because of this reason, biodiesel is essential for enhancing complete combustion, it has a lower calorific value (CV) when contrasted with diesel [1–9].

According to Mahmud et al. [10], palm oil is a potential biofuel source to be used as a fuel in compression ignition (CI) engine. It is the most efficient oil-bearing crop in efficiency, productivity, availability, and land utilization. It produces a higher yield than all other vegetable oils, so it is available at a competitive price. Because of these parameters, palm biodiesel as fuel has been selected in the current research work. The major biodiesel issues were low heating value, low oxidation stability, high brake specific fuel consumption (BSFC), and high NO<sub>x</sub> emissions compared to diesel; to avoid these problems different additives were used. Nano-additives consist of high surface to volume ratio, high calorific value, and high thermal conductivity, So it enhances the performance and emissions of diesel engines. Antioxidant

http://www.ijrtsm.com@International Journal of Recent Technology Science & Management

RESEARCHERID

### () THOMSON REUTERS [Amit et al., 8(8), Aug 2023]

# ISSN : 2455-9679 SJIF Impact Factor : 6.008

additives in biodiesel effectively reduce the free radicals formation and improve the oxidation stability of the fuel. The  $TiO_2$  nano-additive is eco-friendly, low cost, and non-toxic compared to other nano-additives found in the literature. From the literature review, N-phenyl-1,4-phenylenediamine (NPPD) antioxidant additive appears to effectively reduce the NO<sub>x</sub> emissions and enhance the diesel engine's performance compared to other antioxidants.

### **II. OBJECTIVE OF CURRENT STUDY**

The bio-fuels that are most commonly discussed in national and international perspectives are reviewed in his report. The aim of the report is to provide an overview of trends, use and potentials for bio fuels. A distinction is made between those bio fuels that are commercially available today (first generation) and those that will be introduced in the future (second generation). First, a general overview of the fuels' production processes and properties is given, and then of production costs and lifecycle perspectives. Effects of bio fuels on the environment are discussed and also use and potentials in Sweden, Europe and in the world.

Sweden's national goal 2003 concerning the use of biofuels was 3 percent of all fuels in the road transport sector. The actual use reached 2.5 percent. Within the EU the goal is set to 5.75 percent by the end of 2010.

The overview of official documents and reports shows that the future beyond 2020 is relatively uncertain with respect to use of bio-fuels. Reasons include the difficulty in assessing production costs for some fuels that are not commercially available. Also, the distribution of gaseous bio-fuels such as hydrogen is seen as an obstacle for further development.

The current production of bio-fuels in Sweden has a limited potential: about 2 percent of the energy demand required by the road transport sector can be supplied by ethanol from grain. The potential will increase substantially if wood feedstock is used commercially for bio-fuels in the future. An additional 2 percent can be provided by biogas and biodiesel. The aim to use 5.75 percent bio-fuels by 2010 will therefore be hard to fulfill with Swedish resources. Other possibilities, such as aspects of imported ethanol and increased admixture of ethanol in gasoline, are discussed. The energy use of the Swedish vehicle fleet is described in connection with how goals concerning emissions of green house gases can be met.

Emissions from diesel-powered motor vehicles have substantially changed over the last decade as engine manufacturers have significantly lowered the emissions of particles in diesel exhaust through improved engine design and emission-control technologies--in combination with new fuels. The objective of this project is to develop average emission rates of air toxics and other compounds of interest for diesel vehicles using a previously developed database of diesel emissions data for diesel vehicles from 2014 and earlier. These resulting diesel emission rates can then be compared with results of emissions tests on newer diesel vehicles to determine how these engine and fuel changes have affected emissions of air toxics and other compounds.

Air pollution is the presence of pollutants in the atmosphere from man- made or natural substances in quantities likely to harm human, plant, or animal life; to damage man-made materials and structures; to bring about changes in weather or climate; or to interfere with the enjoyment of life or property (Elsom, 1987). The amount of pollutants released to the atmosphere by fixed or mobile man-made sources is generally associated with the level of economic activity. Meteorological and topographical conditions affect dispersion and transport of these pollutants, which can result in ambient concentrations that may harm people, structures, and the environment. In general, the effects on people are most intense in large urban centreswith significant emission sources, unfavourable dispersion characteristics, and high population densities.

Arjun B.Chhetri et al. studied biodiesel (ethyl ester) from waste cooking oil collected from a local restaurant in Halifax, Nova Scotia, Canada. The viscosity of the biodiesel ethyl ester was found to be 5.03 mm 2/sec at 40oC. The viscosity of waste cooking oil measured in room temperature (at 21°C) was 72 mm 2/sec. From the tests, the flash point was found to be 164oC, the phosphorous content was 2 ppm, those of calcium and magnesium were 1 ppm combined, water and sediment was 0 %, sulfur content was 2 ppm, total acid number was 0.29 mg KOH/g, cetane index was 61, cloud point was -10C and pour point was -16oC [14].

#### http://www.ijrtsm.com@International Journal of Recent Technology Science & Management

RESEARCHERID

THOMSON REUTERS

### [Amit et al., 8(8), Aug 2023]

# ISSN : 2455-9679 SJIF Impact Factor : 6.008

Y.Zhang et al. analyzed;Siti Kartina et al. studied waste cooking oil (WCO) is the cheapest source and can reduce problems on waste oil disposal whereas palm fatty acid distillate (PFAD) is byproduct from palm oil refining; therefore can be a readily available feedstock. In this study, two step processes, which combined acidic and alkaline process, were used to convert waste cooking oil and palm fatty acid distillate to biodiesel. It results better gas emissions than the conventional petroleum diesel. There was an average of decrease for carbon dioxide, carbon monoxide and smoke particulate when using biodiesel. A reduction of nearly 80 percent of net carbon dioxide emission was gained when using B100 (neat biodiesel) whereas about 20 percent reduction was achieved when using B20 (20% biodiesel blend [16].

Wei Su et al. studied the food waste utilize in to produce fuel. Base catalyzator was chosen to produce biodiesel from waste oil separated from food waste. The average biodiesel yield could be 85% of waste oil, which occupied 5-10% of food waste. The maximum ethanol concentration of 44 g/L could be acquired from the solid part of food waste by the utilization of yeast. The biodiesel an ethanol production from food waste could reduce waste pollution as well as enhance the utilization value of the waste [17].

B. Nas et al. resulted the biodiesel from waste cooking oil is closer to being cost competitive with petroleum diesel than biodiesel from soybean oil. The use of biodiesel as a fuel it reduces the emissions of particulates and unburned hydrocarbons. Biodiesel reduces nearly all forms of air pollution. Most importantly, biodiesel reduces air toxics and cancer-causing compounds. There is no sulfur in biodiesel, so biodiesel will not contribute to sulpher dioxide emissions or poison exhaust catalysts. B20 has 20% of the benefits of pure biodiesel. B20 can also reduce the soot and smell of diesel exhaust [18].Biodiesel has a viscosity much closer to diesel fuel than vegetable oil.[19]

B. Supplea et al. analyzed the effect of steam injection and sedimentation treatment of waste cooking oil on the quality of TG, to be used as a raw material for the production of biodiesel, was investigated. The effect of steam treatment was evaluated in terms of a number of physical and chemical characteristics.Significant decreases in the moisture, FFA and PV. The decreases in moisture from 1.4 to 0.4% and in FFA from 6.27 to 4.28% were found to correlate strongly with an increase in yield of ester from 67.5 to 83.5%. There was a significant change in the physical properties of the oil (e.g., density and kinematic viscosity) with treatment. Then use of biodiesel as a fuel produces low emissions of carbon monoxide, particulate matter and unburned hydrocarbons [20].

### III. PROCEDURES, PREPARATION OF BIODIESEL FROM KARANJA OIL

#### **Biodiesel Production**

The production of biodiesel, alkyl ester is well known. There are three basic routes to ester production from oils and fats:

- Based catalyzed transesterification of the oil with alcohol
- Direct acid catalyzed etherification of the oil with methanol
- Conversion of oil to fatty acids, and then to alkyl ester with acid catalysis.

### **IV. CONCLUSION**

In this study, experiments were conducted with a direct injection, naturally aspirated diesel engine to investigate the performance and emission of the biodiesel and its blends prepared from Karanja oil. This chapter describes the conclusions for Performance and unregulated emission of diesel and biodiesel fuelled engine. So finally biodiesel blends few increase to for It is concluded that the use of biodiesel blends slightly increases the Brake Specific Fuel Consumption in comparison to use of fuel of diesel as same loading condition. The BTE of biodiesel fuel were finding out to be comparable as that of diesel fuel when the blends were 20 %. The emissions of biodiesel fuel is better compare to diesel.

# RESEARCHERID THOMSON REUTERS [Amit et al., 8(8), Aug 2023]

#### REFERENCES

- [1] Ban-Weiss, G. A., Chen, J. Y., Buchholz, B. A., and Dibble, R. W. 2007. A numerical investigation into the anomalous slight NOx increase when burning biodiesel: A new (old) theory. Fuel Process. Technol. 88:659–667.
- [2] Vicente, G., Martinez, M., and Aracil, J., —Integrated biodiesel production: a comparison of different homogenous catalysts systems, Bioresource Technology, 92, 2004, pp: 297–305.
- [3] Sharma, Y.C. and Singh, B., —Development of biodiesel: current scenario, Renewable Sustainable Energy Reviews, 13, 2009, pp: 1646–1651.
- [4] Freedman, B.E., Pryde, H., and Mounts, T.L., —Variables affecting the yields of fatty esters from transesterified vegetable oils, JAOCS, 61, 1984, pp: 1638–1643.
- [5] Vicente, G., et al., —Application of the factorial design of experiments and response surface methodology to optimize biodiesel production, I Industrial Crop Production, 8, 1998, pp: 29–35.
- [6] Feuge RO and Gros AT, 1949, Modification of vegetable oils, VII, Alkali catalyzed in teresterification of peanut oil with ethanol, Journal of the American Oil Chemists' Society, 26: 97-103.
- [7] Chowdhury DK and Mukherji BK, 1954, Ethanolysis of vegetable oils using sodium hydroxide as catalyst, II, Effect of temperature and moisture, Journal of the Indian Chemical Society, 31(125-128).
- [8] Gauglitz EH and Lehman LW, 1963, The preparation of alkyl esters from highly unsaturated triglycerides, Journal of the American Oil Chemists' Society, 40: 197-198. [24] Glass RL, 1963, Alcoholysis, saponification and the preparation of fatty acid methyl esters, Lipids, 6:919-925.
- [9] Freedman, B.E., Pryde, H., and Mounts, T.L., —Variables affecting the yields of fatty esters from transesterified vegetable oils, JAOCS, 61, 1984, pp: 1638–1643.
- [10] Szpiz RR, Jablonka FH, Pereira DA, Lago RCA and Hartman L, 1984, Processo datransesterificacao de oleos vegetais, BR8300429 (patent)